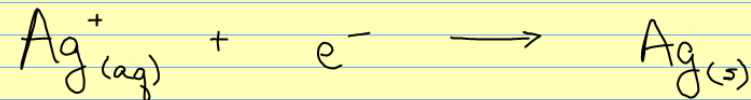
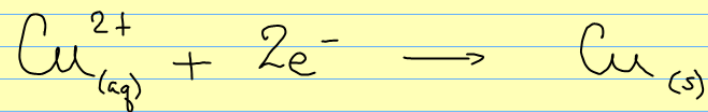


Unit 4: Electrochemistry.

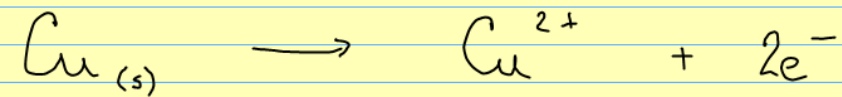
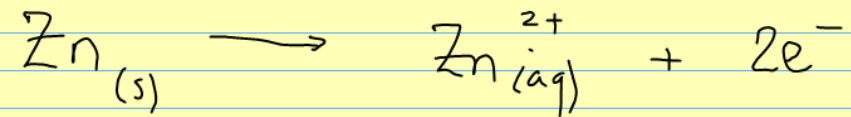
May 25, 2011.

Redox Reactions

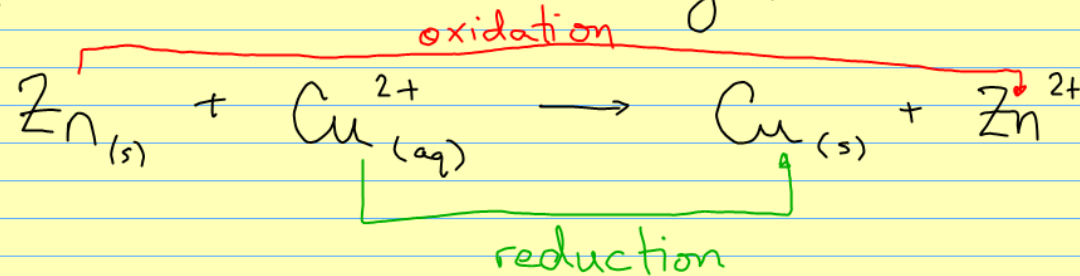
A redox reaction is a combination of a reduction and an oxidation reaction. Reduction is a process in which an atom or ion gains electrons.

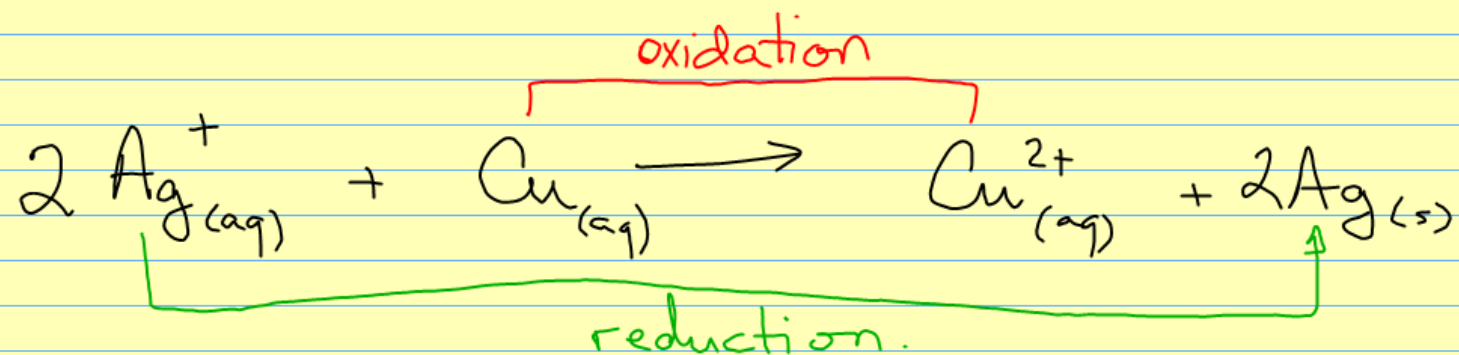


Oxidation is the process in which an atom or ion loses electrons.



In a redox reaction reduction and oxidation occur together





Oxidation
Is
Loss

Loss
Electrons
Oxidation

Reduction
Is
Gain

Gain
Electrons
Reduction.

P 376 Practice.